

#### **RESEARCH ARTICLE**

# Compounds with neuroprotective activity from the medicinal plant *Machilus thunbergii*

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#### **Abstract**

The dichloromethane fraction of the bark of *Machilus thunbergii* Sieb. et Zucc. (Lauraceae) significantly protected primary cultures of rat cortical cells exposed to the excitotoxic amino acid, L-glutamate. Through the activity-guided isolation from the  $CH_2Cl_2$  fraction, (+)-9'-hydroxygalbelgin (1), isogalcatin B (2), (75,85,8'R)-3',4'-dimethoxy-3,4,-methylenedioxylignan-7-ol (3), 1-hydroxy-7-hydroxymethyl-6-methoxyxanthone (4), 5,7-dimethoxy-3',4'-methylenedioxyflavan-3-ol (5), (+)-(35,45,6R)-3,6-dihydroxypiperitone (6), protocatechuic acid methyl ester (7) and tyrosol (8) were obtained. All of them had significant neuroprotective activities against glutamate-induced neurotoxicity in primary cultures of rat cortical cells at concentrations ranging from 0.1  $\mu$ M to 10.0  $\mu$ M and were comparable to MK-801, a well-known inhibitor of glutamate receptor.

**Keywords:** Machilus thunbergii; Tyrosol; Neuroprotective; Antioxidant

## Introduction

Glutamate is an excitatory amino acid and activates different types of ion channel-forming receptors and G-proteincoupled receptors. In the central nervous system, glutamate plays its essential roles such as neuronal survival, synaptogenesis, neuronal plasticity, memory, learning and behavior [1, 2]. However, one of the pathogenetic mechanisms discussed to be relevant for the etiology of Alzheimer's disease (AD) are the glutamate induced excitotoxic cascades [3]. Thus, glutamate excitotoxicity has also become a target for AD treatment. We previously applied glutamate induced neurotoxicity in primary cultures of rat cortical cells as an in vitro assay system to isolate neuroprotective compounds from natural products [4]. The methanolic extract of the bark of Machilus thunbergii Sieb. et Zucc. (Lauraceae) was found to significantly protect primary cultures of rat cortical cells against glutamate-induced neurotoxicity.

*Machilus thunbergii* (Lauraceae) is widely distributed in Korea. The cortex of *M. thunbergii*, which has been consumed as traditional herbal medicine for a long period of time, has provided Korean farmers with significant income. The cortex are used in Korean folk medicine for treatment of leg edema and abdominal distension and pain [5]. Many

lignans, flavonoids and essential oils have all been previously reported as components of the bark of M. thunbergii [6]. In our previous study, subsequent phytochemical studies coupled with our bioassay were performed to elucidate active principles responsible for the neuroprotective activity of M. thunbergii. We reported that meso-dihydroguaiaretic acid, licarin A, isoguaiacin and guaiacin isolated from M. thunbergii had the significant neuroprotective activity against glutamate-induced neurotoxicity [7]. In addition to lignans such as meso-dihydroguaiaretic acid, licarin A, isoguaiacin and guaiacin, we attempted to isolate more neuroprotective compounds from the  $CH_2Cl_2$  fraction of M. thunbergii methanolic extract against glutamate-induced neurotoxicity as measured  $in\ vitro$ .

## Materials and methods

## General experimental procedures

The <sup>1</sup>H- and <sup>13</sup>C-NMR measurements were carried out in a Bruker AMX 400 spectrometer operating at 400 and 100 MHz, respectively. TMS or solvent signals were used as internal standard. FT-IR spectra were recorded on a Perkin-Elmer 1710 spectrophotometer. UV spectra were recorded

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on a Shimadzu UV-2100 spectrophotometer. EIMS spectra were obtained on a VG Trio II spectrometer. Column chromatography was performed on Merck (9025) silica gel 60 (0.04-0.063 mm). Analytical TLC was performed on precoated Merck  $\rm F_{254}$  silica gel plates and visualized by spraying with anisaldehyde- $\rm H_2SO_4$ . An HPLC system (Hitachi L-6200, Japan) equipped with a UV-visible detector and Microsorb  $\rm C_{18}$  semipreparative column (Rainin Inst. Co.) was used for isolation. CD data were recorded in MeOH on a JASCO-J715 spectrophotometer.

#### Plant materials

The bark of *M. thunbergii* Sieb. et Zucc. was purchased in a local market for Oriental medicine in Gyeongdong, Seoul, Korea, in 2001. Voucher specimens (SNUPH-0521) have been deposited in the Herbarium of the College of Pharmacy, Seoul National University.

## Extraction and isolation

The dried bark of M. thunbergii (30kg) was ground into a powder and extracted 3 times with 80% MeOH using the reflux apparatus. Upon removal of the solvent in vacuo, the methanolic extract yielded 2.9kg of material (9.7% by dry weight). The methanolic extract was suspended in H<sub>2</sub>O and partitioned successively with CH2Cl2. The CH2Cl2 fraction (980g) which showed significant neuroprotective activity was subjected to column chromatography (cc) over silica gel (12  $\times$  80 cm) eluted with *n*-hexane-EtOAc (20:1, 10:1, 5:1, 2:1, 1:1, EtOAc, MeOH) to afford seven fractions (F01-F07). F03 was further applied to a silicagel cc  $(3.5 \times 45 \text{ cm})$  eluting with n-hexane-EtOAc (20:1, 10:1, 5:1, 2:1, 1:1, EtOAc, MeOH), affording ten fractions (F03-I to F03-X). F03-V was subjected to a Sephadex LH-20 column eluting with MeOH to afford seven fractions (F03-V-01 to F03-V-07). Among the 7 subfractions, F03-V-04 (125 mg) yielded compound 3 (3.3 mg) and 4 (13.2 mg) by additional purification steps on RP-18 HPLC (H<sub>2</sub>O-AcCN, 65:35). F04 was further applied to a silica gel cc  $(3.5 \times 45 \,\mathrm{cm})$  eluting with *n*-hexane-EtOAc (20:1, 10:1, 5:1, 2:1, 1:1, EtOAc, MeOH), affording three fractions (F04-I to F04-III). Among the 3 subfractions, F04-III (85 mg) yielded compound 1 (3.6 mg) by additional purification steps on RP-18 HPLC (H<sub>2</sub>O-AcCN, 60:40) and compound **5** (6.3 mg). F05 was further applied to a silica gel cc  $(3.5 \times 45 \text{ cm})$  eluting with *n*-hexane-EtOAc (20:1, 10:1, 5:1, 2:1, 1:1, EtOAc, MeOH), affording five fractions (F05-I to F05-V). From the subfraction F05-II, compounds 2 (3.4 mg), 6 (35.7 mg), 7 (5.0 mg) and 8 (2.1 mg) were isolated by additional purification steps on RP-18 HPLC (H<sub>2</sub>O-AcCN, 60:40). Optical rotations were determined on a polarimeter at 25°C.

(+)-9'-hydroxygalbelgin (1): White amorphous powder. [α]<sup>25</sup><sub>D</sub> +19.0° (CHCl<sub>3</sub>, 2.44). IR (KBr)  $v_{\text{max}}$ : 3440, 1510, 1450, 1271, 1118 cm<sup>-1</sup>. <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>) δ: 1.08 (3H, d, J= 6.6 Hz, H-9), 1.99-2.07 (1H, m, H-8), 2.25-2.35 (1H, m, H-8'), 3.71 (2H, d, J= 7.1 Hz, H-9'), 4.18 (1H, d, J= 9.0 Hz, H-7), 4.59 (1H, d, J= 8.9 Hz, H-7'), 5.93 (2H, s, OCH<sub>2</sub>O), 5.94 (2H, s, OCH<sub>2</sub>O), 6.73 (1H, d, J= 8.1 Hz, H-5), 6.74 (1H, d, J= 8.0 Hz, H-5'), 6.76 (1H, dd, J= 1.7, 8.0 Hz,

H-6), 6.77 (1H, dd, J = 1.7, 8.0 Hz, H-6′), 6.85 (1H, d, J = 0.8 Hz, H-2′) ppm,  $^{13}$ C-NMR (100 MHz, CDCl<sub>3</sub>)δ: 16.8 (C-9), 46.3 (C-8), 54.0 (C-8′), 70.0 (C-9′), 77.2 (C-7′), 89.4 (C-7), 100.9 (OCH<sub>2</sub>O), 101.1 (OCH<sub>2</sub>O), 106.7 (C-2′), 106.7 (C-2), 107.9 (C-5′), 108.1 (C-5), 119.9 (C-6), 120.0 (C-6′), 134.9 (C-1′), 137.3 (C-1), 147.1 (C-4′), 147.3 (C-4), 147.8 (C-3′), 147.9 (C-3) ppm. EIMS m/z (rel int) 356 [M]<sup>+</sup> (52%), 255 (44%), 250 (22%), 208 (87%), 192 (70%), 173 (100%). HREIMS m/z 356.1258 (calcd for  $C_{20}H_{20}O_6$ : 356.1259).

**isogalcatin B** (2): White amorphous powder.  $[\alpha]^{25}$ -21.1° (CHCl<sub>3</sub>, 1.25). IR (KBr)  $v_{\rm max}$ : 2900, 1610, 1510, 1495, 1450 cm<sup>-1</sup>. <sup>1</sup>H-NMR (400 MHz, CDCl<sub>2</sub>) δ: 0.89 (3H, d, J = 6.7 Hz, H-9, 0.90 (3H, d, J = 6.9 Hz, H-9'), 1.88-1.91 (1H, m, H-8'), 2.00-2.02 (1H, m, H-8), 2.45 (1H, dd, J = 8.1, 16.5 Hz, H-7a), 2.84 (1H, dd, J = 5.4, 16.5 Hz, H-7b), 3.66 (1H, d,  $I = 5.6 \,\mathrm{Hz}$ , H-7'), 3.68 (3H, s, OCH<sub>2</sub>), 3.86 (3H, s, OCH<sub>2</sub>), 5.91  $(2H, s, OCH_2O)$ , 6.33 (1H, s, H-5), 6.48 (1H, dd, J=1.5, 8.1)Hz, H-6'), 6.49 (1H, d, J = 1.5 Hz, H-2'), 6.58 (1H, s, H-2), 6.70 (1H, d, J = 8.3 Hz, H-5'), <sup>13</sup>C-NMR (100 MHz, CDCl<sub>2</sub>) δ: 15.3 (C-9), 16.6 (C-9'), 28.4 (C-8), 34.6 (C-7), 40.9 (C-8'), 51.0 (C-7'), 55.7 (OCH<sub>3</sub>×2), 55.8 (OCH<sub>3</sub>×2), 100.8 (OCH<sub>2</sub>O), 107.6 (C-2), 109.4 (C-2'), 111.2 (C-5'), 113.2 (C-5), 122.1 (C-6'), 128.4 (C-1), 129.3 (C-6), 141.3 (C-1'), 145.5 (C-4), 147.1 (C-4'), 147.3 (C-3), 147.4 (C-3'). EIMS m/z (rel int) : 340 [M]<sup>+</sup> (34%), 309 (41%), 283 (64%), 269 (37%), 253 (100%), 223 (76%), 187 (65%). HREIMS m/z 340.1672 (calcd for  $C_{21}H_{24}O_4$ : 340.1675).

## In vitro neuroprotective activity

Female Sprague-Dawley rats (20-23 °C; 12 h light cycle from 09:00 to 21:00; food, Agribrand Purinar Korea, and water ad libitum) were provided by the Laboratory Animal Center, Seoul National University. All experiments were conducted according to the guidelines of the Committee on Care and Use of Laboratory Animals of the Seoul National University. Primary cultures of mixed cortical cells containing both neuronal and glial cells were prepared from late fetal SD rats (17-19-days gestation in utero) as described previously [8]. All compounds were dissolved in DMSO (final culture concentration, 0.1 %). Preliminary studies indicated that the solvent had no effect on cell viability at the concentration used (data not shown). Two known glutamate receptor antagonists, MK-801 (dizocilpine maleate, a noncompetitive antagonist of the NMDA receptor) and CNQX (6-cyano-7-nitroquinoxaline-2,3-dione, a non-NMDA receptor antagonist) were used as positive controls for the assessment of neuroprotective activity [8, 9]. Cortical cell cultures were washed with DMEM and incubated with the test compounds for 1 h. The cultures were then exposed to 100 µM glutamate. After 24 h incubation, the cultures were assessed for the extent of neuronal damage. Neuronal viability and integrity were quantified by MTT (3-[4,5dimethylthiazol-2-yl]-2,5-diphenyltetrazolium bromide) and/or LDH (lactate dehydrogenase) assay described in our previous report [8]. Data are expressed as the percentage protection relative to control cultures.

Figure 1. Structures of compounds 1-8 isolated from the CH<sub>2</sub>Cl<sub>2</sub> fraction of *M. thunbergii* 

## Statistical analysis

Data were evaluated for statistical significance by "ANOVA" test using a computerized statistical package (control vs 0.1  $\mu M$ , control vs 1.0  $\mu M$  and control vs 10.0  $\mu M$ , respectively).

We performed all the statistics on the raw data prior to transformation to percentage against control. The confidence level for statistical significance was set at a probability value of 0.05.

# Results and discussion

As a part of our continuing research seeking neuroprotective compounds from natural resources, the methanolic extract of the bark of Machilus thunbergii Sieb. et Zucc. (Lauraceae) was found to significantly protect primary cultures of rat cortical cells against glutamate-induced neurotoxicity. We attempted to isolate more neuroprotective compounds from the CH<sub>2</sub>Cl<sub>2</sub> fraction of M. thunbergii methanolic extract against glutamate-induced neurotoxicity as measured in vitro. As a result, we isolated and identified two new lignans, (+)-9'-hydroxygalbelgin (1) and isogalcatin B (2), and six known compounds (Figure 1). The methanolic extract of bark of M. thunbergii was found to exhibit a significant neuroprotective activity against glutamate-induced toxicity. The methanolic extract was suspended in H<sub>2</sub>O and partitioned with CH<sub>2</sub>Cl<sub>2</sub>. At a concentration of 10 µg/mL, neuroprotection against glutamateinduced toxicity of CH<sub>2</sub>Cl<sub>2</sub> fraction was 65.7 % (p<0.01). Further fractionation and separation of the CH<sub>a</sub>Cl<sub>a</sub> fraction by several chromatographic methods yielded eight compounds. Spectral data of compounds 3-8 matched those of (7S,8S,8'R)-3',4'-dimethoxy-3,4,-methylenedioxylignan-7-ol (3), 1-hydroxy-7-hydroxymethyl-6-methoxyanthone (4), 5,7-dimethoxy-3',4'-methylenedioxyflavan-3-ol (5), (+)-(3S,4S,6R)-3,6-dihydroxypiperitone (6), protocatechuic acid methyl ester (7) and tyrosol (8) [10–16].

Compound 1 was obtained as amorphous powders. The molecular formula  $C_{20}H_{20}O_6$  was established by HREIMS, m/z 356.1258 (calcd for  $C_{20}H_{20}O_6$ : 356.1259) and <sup>13</sup>C NMR data. It contained two 3,4-methyenedioxyphenyl groups as established by the presence of two sets of ABX system for six aryl protons (one for H-2, H-5 and H-6, and the other for H-2', H-5' and H-6') and two metylenedioxy singlets ( $\delta$  6.04 and 5.94) in its <sup>1</sup>H NMR spectrum. The spectral data indicated 1 contained two relatively deshielded benzylic protons ( $\delta$  4.18 (1H, d, J = 9.0 Hz, H-7) and  $\delta 4.59 (1H, d, J = 8.9 Hz, H-7')$ , two aliphatic methanes ( $\delta$  1.99-2.07 (1H, m, H-8),  $\delta$  2.25-2.35 (1H, m, H-8')), and two protons bound to hydroxyl group ( $\delta$ 3.71 (2H, d, J = 7.1 Hz, H-9')). On the basis of these data, 1 was inferred as 9'-hydroxy-3,4:3',4'-bis(methylenedioxy)-7,7'-epoxylignan [17]. In the NOESY spectrum, H-7 and H-8' peaks correlated each other, and H-7' and H-8 peaks correlated each other. We determined trans H-7/H-8, trans H-7/H-7', trans H-7'/H-8', and trans H-8'/H-8 relationships. Absolute configuration was determined by comparison of CD data of compound 1 and (+)-galbelgin, which has the similar stereochemistry of compound 1 (Table 1) [18, 19]. From the spectroscopic data above, the structure of 1 was concluded to be (+)-9'-hydroxygalbelgin.

Compound **2** was obtained as white amorphous powders. The molecular formula  $C_{21}H_{24}O_4$  was established by HREIMS, m/z 340.1672 (calcd for  $C_{21}H_{24}O_4$ : 340.1675) and <sup>13</sup>C NMR data. The spectral data were very similar to those of (-)-isoguaiacin, except for the presence of a methylenedioxy group ( $\delta$  5.91 (2H, s, OCH<sub>2</sub>O)) and the position of a methoxy group ( $\delta$  3.68 (3H, s, OCH<sub>3</sub>)) [6]. The relative configuration of chiral

Table 1. CD data of compound 1 and (+)-galbelgin

	Δε		
	Compound 1 (c=1.0mg/mL)	(+)-galbelgin ( $c$ =0.8mg/mL)	
312	+0.11	+0.09	
304	+11.55	+8.05	
290	-3.24	-1.89	
269	+3.69	+2.79	
260	+0.54	+0.41	
255	+12.51	+7.14	
233	+2.67	+1.82	
223	+3.84	+2.87	
206	+4.83	+4.15	
204	+5.67	+5.44	

**Table 2.** The Protective activity of compounds isolated from the  $\mathrm{CH_2Cl_2}$  fraction of M. thunbergii bark against glutamate-induced neurotoxicity in primary cultures of rat cortical cells

	Protection (%)		
	0.1 μΜ	1.0 μΜ	10.0 μΜ
control		$100.0 \pm 0.21$	
glutamate-treated <sup>a,b</sup>		$\boldsymbol{0.0 \pm 0.04}$	
1	$16.8 \pm 1.64$	$25.4 \pm 2.22$	$30.4 \pm 2.66^*$
2	$15.8 \pm 2.11$	$20.1\pm1.80$	$30.8\pm4.24^{^{\ast}}$
3	$21.4\pm2.64^{^{\ast}}$	$23.5\pm2.19^{^{\ast}}$	$25.1\pm2.15^{^{\ast}}$
4	$18.4 \pm 2.84$	$19.7 \pm 3.64$	$27.6\pm2.88^{^{\ast}}$
5	$25.1\pm2.55^{^{\ast}}$	$28.8\pm4.51^{^{\ast}}$	$31.1 \pm 1.22^{**}$
6	$10.1\pm1.10$	$20.4\pm2.19$	$28.1\pm3.25^{^{\ast}}$
7	$30.0 \pm 2.58^*$	$46.2 \pm 2.54^{**}$	$50.5 \pm 3.66^{**}$
8	$29.4 \pm 2.11^*$	$55.5 \pm 2.88^{***}$	$70.2 \pm 3.69^{***}$
MK-801 <sup>c</sup>	$51.7 \pm 4.51^{***}$	$68.8 \pm 3.88^{***}$	$79.3 \pm 2.01^{***}$
$CNQX^d$	$48.8 \pm 2.14^{***}$	$57.9 \pm 2.41^{***}$	$69.5 \pm 1.53^{***}$

 $^{\rm a}\text{LDH}$  released from control and glutamate-treated cultures was  $109.5\pm7.4$  and  $196.3\pm8.3$  mU/mL, respectively. Cell viability was calculated as  $100\times(\text{LDH}$  released from glutamate-treated - LDH released from glutamate + test compound) / (LDH released from glutamate-treated - LDH released from control).  $^{\rm b}\text{Glutamate}$ -treated value differs significantly from the untreated. Control at a level of p<0.001.  $^{\rm c}\text{MK-}801$ : dizocilpine maleate, a non-competitive antagonist of the NMDA receptor.  $^{\rm d}\text{CNQX}$ : 6-cyano-7-nitroquinoxaline-2,3-dione, non-NMDA receptor antagonist.The values are expressed as means  $\pm$  SD of triplicate experiments. Results differ significantly from the glutamate-treated.  $^{\rm t}\text{p}<0.05, \,^{\rm t}\text{p}<0.01, \,^{\rm tt}\text{p}<0.001$ 

protons was determined as *cis* H-8/H-8′ and *trans* H-7′/H-8′ relationships by comparison to  $^{13}$ C NMR data of reference [20]. The position of methoxy group was also determined by comparison to reference [21]. Optical rotation was determined on polarimeter at 25°C. The optical rotation value of the compound **2** was -21.1° (c 1.25 in CHCl<sub>3</sub>). From the spectroscopic data above, the structure of **2** was designed to be isogalcatin B as a stereoisomer of isogalcatin.

Compounds 1-8 were tested for protective activity against glutamate-induced toxicity. The relative protection

of the eight compounds is compared in Table 2. All of these comounds showed significant neuroprotective activities at concentrations ranging from  $0.1\mu M$  to  $10.0~\mu M$  (Table 2; MTT assay showed same trend of LDH assay; data not shown). Of the four neuroprotective compounds, tyrosol (8) showed the most potent activity against glutamate-induced neurotoxicity. The potency of the tyrosol is similar to that of MK-801 or CNQX, the positive controls.

Tyrosol is a major compound of extra virgin olive oil. Extra virgin olive oil, the typical added fat of the Mediterranean diet, has been related to a general health benefit and a reduced incidence of risk factors for coronary heart disease [22]. Tyrosol inhibited cell-mediated oxidation of LDL and preserved cellular activities of antioxidative enzymes such as glutathione peroxidase and glutathione reductase reduced in the in J774 A.1 cells incubated with LDL. And tyrosol also was also effective at preserving the GSH content completely. Tyrosol inhibited the ROS production in J774 A.1 cells during cell mediated oxidation of LDL [23]. Defections in GSH metabolism might cause oxidative stress, which has been implicated in several neurologic and neurodegenerative diseases [24].

At present, the cellular and molecular mechanisms that underlie the action of tyrosol are not fully understood. However, our result and other previous reports indicate that tyrosol significantly protects primary cultured neuronal cells against glutamate-induced oxidative stress via antioxidative activities. Therefore, we conclude that tyrosol might offer useful therapeutic choice in treatment of neurodegenerative disorders caused by excitotoxicity.

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